An Insight into Removal and Recovery of SO₂ and CO₂ in Petroleum Industries with Emphasis on Absorption

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Abstract: The exhaust gases from petroleum refineries contain carbon dioxide, sulphur dioxide and oxides of other compounds. Methods such as adsorption, absorption, membrane separation, chemical conversion and electrostatic separators can be used for the removal of these gases from the gaseous mixtures. The recovery of the valuable gases, which can be reused in the processes, is also important aspect of waste gas treatment. Membrane related treatment methods are effective but high cost limits their use to specific applications. Adsorption is also effective in some applications. Regeneration and solid disposal is inconvenient. Absorption is most widely used method for recovery of gases. Current review summarizes research and studies on absorption for removal and recovery of carbon dioxide and sulphur dioxide in petroleum industry waste gases.

Keywords: solvent, recovery, capture, cost, regeneration.

1. INTRODUCTION

The waste gases from refineries contain oxides of sulphur and carbon along with oxides of nitrogen. Sulphur dioxides concentration in air needs to be maintained below the limits prescribed by the regulatory authorities and also from human health point of view. The air pollution control includes removal of particulate matter and various gases from the waste gases. Cyclone separators, electrostatic separators and bag filters can be used for separation of particulate matter from the flue gases [1-5]. Removal of gases like carbon dioxide and sulphur dioxide can be carried out by methods like adsorption and absorption [6-11]. The current review is aimed at summarizing research and studies on sulphur dioxide and carbon dioxide removal from waste gases of petroleum refineries and industries with emphasis on absorption.

2. An Insight into Removal and Recovery of SO_2 and CO_2 in Petroleum Industries with Emphasis on Absorption

Mandal and Krishna conducted experiments to find the better suitable amine solutions to facilitate the maximum SO_2 absorption from flue gases [12]. In their investigation, they also checked the possible solvent degradation. They conducted experiments to check the effect of CO_2 present in effluent while absorption taking place. They observed that the rate of absorption was more in absence of CO_2 for two solvents. For other solvents there was no effect of CO2.Songolzadeh et.al. Carried out review on carbon dioxide removal from flue gases [13]. According to them, carbon capture and storage (CCS) is a major strategy that can be used to reduce GHGs emission. They studied various technologies for CO_2 capture including absorption, adsorption, cryogenic distillation, and membrane separation. According to these studies, flue gases properties (mainly concentration of CO_2 , temperature and pressure) are the most effective factors for selection of suitable process for CO₂separation. According to these studies membrane and adsorption processes need to be developed further in terms of simulation and modeling. Chapel et.al. Studied commercial trends in removal and recovery of CO_2 from flue gases [14]. According to them, the interest of investigators in CO_2 recovery is driven by factors like the merchant CO_2 market, renewed interest in enhanced oil recovery (EOR), and the desire to reduce greenhouse gas emissions. They also discussed key process design and operating issues for amine chemical solvent CO₂ recovery processes. Also they summarized the competitive processes for CO_2 recovery from flue gases. The recovery cost varies with the plant size, the market price of CO_2 , the flue gas source, and government regulation. The results of these experimentations revealed that chemical absorption process produces the least costly CO₂.

Zare and Anvaripour discussed removal of SO_2 and NO_x from flue gas [15]. In their studies, they used corona discharge to remove SO_2 and NOx. They connected to an AC power supply with a highvoltage transformer. In their investigation, they observed that as the discharge power of the plasma increased the removal efficiency of three pollutants increased. Also it was shown that at gap spacing of 3 to 5 mm the removal efficiency increased to a maximum. Wang and Han reviewed CO₂ capture with polymeric Membrane [16]. According to them, hollow fiber membrane is highly gas permeable. The problem with these membranes is that they are not selective. Solvent helps in realizing selectivity in the membranes. According to them, Pressure and vacuum combination will accelerate the mass transfer rate, reasonable cost. They also concluded that membrane length can be increased to increase separation efficiency. Parallel membrane devices will also increase the separation efficiency. Rahmandoost et.al. carried out studies and experimentation on CO₂ recovery from flue gases[17]. They presented the results of studies on chemical methods of absorbing CO₂ from flue gases with the new solvent. In their investigation, they also evaluated the effects of operating conditions on CO_2 absorption efficiency. They observed that the CO_2 absorption rate in new solvent was higher than conventional solvent (monoethanolamine). They observed that the CO₂ absorption efficiency increases with the concentration of absorbent. With the absorbent flow rate, also, the CO_2 absorption efficiency increases. It was also observed that with CO₂ concentration, the absorption decreased marginally.

Nagata et.al. carried out experimental studies to reduce SO_3 from flue Gas[18]. Their studies were focused on flue gas emitting from combusting high sulfur content residual materials in oil refinery. Menhali carried out studies on the effects of flue gas recirculation (FGR) ratio and excess air (EA) in a natural gas-fired turbine for an effective CO_2 absorption by amines[19]. In his review, Kulkarni discussed flue gas treatment technologies[20]. According to them CO_2 can be removed by various methods including absorption by chemical solvents, physical absorption, cryogenic separation, membrane separation. The selection of proper method depends on factors such as concentration, cost, resources and characteristics of flue gases.

3. CONCLUSION

The investigations carried out by various researchers indicated that CO_2 can be removed by various methods including absorption by chemical solvents, physical absorption, cryogenic separation, membrane separation. Studies revealed that the selection of proper method depends on factors such as concentration, cost, resources and characteristics of flue gases. Studies indicated that CO_2 absorption efficiency increases with the concentration of absorbent. With the absorbent flow rate, also, the CO_2 absorption efficiency increases. Various investigators have carried out investigations on removal of sulphur dioxides and carbon dioxides from of petroleum industry waste gases. Studies also indicated that the composition of waste gases affected the effectivity of treatment and the absorption is effective for waste gas treatment.

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