Processing of Non-Fermented Sour-Sweet-Slices of Cucumbers (*Cucumis Sativum* L)

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Abstract: This research is concentrated about the effects of prickling, blanching, storage temperature and time focusing on establishment of equilibrium, drained weight and firmness of small sour-sweet cucumber that are used on rye bread and as an ingredient to dinner foods. Drained weight of pickled cucumber increased significantly with storage time up to 114 days whereas drained weight of non-pickled cucumber decreased up to storage for 187 days. Soluble solids and acidity were significantly higher for pickled cucumber in comparison to non-pickled cucumber. Soluble solids and acidity was non-significantly different for pickled and non-pickled cucumber and brine components. Drained weight was significantly with days of storage up to 10 min by blanching at 72 °C and 12 min at 85 °C. Drained weight after blanching at 85°C was 30 g in comparison to 12 g after blanching at 72 °C. Cucumber firmness increased on average 7 kg during blanching at 72 °C min, whereas the firmness decreased 27 kg during blanching at 85 °C. Drained weight increased significantly with storage time, while firmness decreased significantly with storage time. Storage of pickled cucumber at 12 and 20 °C did not result in significances in texture and cucumber weight. Blanching of pickled cucumber at 12 and 20 °C showed that drained weight increased significantly with blanching time and that firmness decreased with increases in the storage temperature.

Keywords: *Slicing - blanching - firmness -enzymes – browning*

1. INTRODUCTION

This research is concentrated about the effects of prickling, blanching and storage temperature and time focusing on establishment of equilibrium, drained weight and firmness of small sour-sweet cucumber.Preserved sour sweet slices of large cucumber produced in brine with sugar, vinegar and some spices are normally used as a decoration and flavour ingredient upon several kinds of open rye bread sandwiches. They are often served with hot dishes together with different kinds of meats, eventually boiled potatoes and other vegetables or spices. The decreasing sale of this kind of cucumber could be a lack in knowledge regarding the primary cucumber properties and missing knowledge among industrial producers about how to optimize the primary processing steps regarding improvement of this delicious food. Size and regularity in form, colour, texture, sweet-sour flavour and storability are very important among the consumers and must also be one major interest for production of delicious and healthy cucumber products[1]. The major interest among horticulturists is data regarding delivering of uniform raw cucumber, storage requirements, rate of deliveries and management. Besides of these areas is knowledge about time, temperature, pasteurization, brine composition and packaging, storage conditions and storage facilities for the cucumber products to be delivered to the super markets. And therefore, it was decided to study the effects of pickling, blanching, packaging, pasteurization, and brine composition in order to optimize the quality of mall non-fermented sour-sweet pickled cucumber.

2. MATERIALS AND METHODS

Production of cucumbers with length 7-12 cm was washed in tap water (12-13 °C). The cucumbers were prickled using a piece of wood with 1.4 mm thick nails in squares with border length 15 mm. That tool was applied for prickling immediately before processing using size of lots with 1300 g. Experiment one was processed using 1300 g non-pickled and 1300 g freshly pickled cucumber that was blanched at 85 °C for 6 min in water baths with 20 L water and then cooled in 20 L cold tap water. The jars (580 cm³) were packed with 315 g blanched cucumber and 265 g brine containing 37 w/w% sugar, 2.2 w/w% acetic acid, 2 g 1^{-1} sodium benzoate and 2 g 1^{-1} potassium sorbate. The jars with cu-

cumber were pasteurized at 85 °C for 15 min in water baths and then cooled in cold tap water. The glass jars were cooled 5 min in water at 60 °C and in cold water at 12 °C and stored in a cooling cabinet at 12 °C. Experiment two included blanching at 72 °C for 0, 3, 6, 10 and 15 min and blanching at 85 °C for 0, 3, 6, 9 and 12 min and storage at 12 °C.Drained weight of cucumber slices was determined after draining for 3 min on a standard sieve ASTM.E aperture 2.8 mm, 11 mesh no 7, diameter 20 cm (Endecott's, England). The cucumbers were cut into circular pieces with thickness 6.8 mm us-Firmness of the stored cucumbers were measured using an ing a Ditocutter (TR, Wycombe, UK). Instron apparatus (Instron Ltd, Food Tester model 5565, High Wycombe, U K) equipped with a Kramer-shear cell and registering the maximum force (kg) to break 75 g sliced cucumber with a sample cell piston velocity 20 cm min⁻¹. Soluble solids were determined using a refractometer (Bellingham and Stanley, RFM 800, Turnbridge, Turnbridge Wells, Kent, UK). Acidity was measured by titration with 0.1 N NaOH (Bie and Berntsen, Rødovre, Denmark) to pH 8.1. Samples of raw, cooked and drained cucumber materials for chemical analyses were obtained by maceration of the drained slices in a minimum of double distilled water or other liquids using a Waring blender (Waring Products B012-34-BL 99, New Hartford, USA). The sensory taste panel included people age 18-28 year interested in sensory evaluation was tested for skills in the basal tastes bitterness, sourness, sweetness, saltiness and a diversity of various food odours were trained intensively in evaluation of the sensory characteristics of at least six different raw or processed fruits and at least six raw or processed vegetable foods, previously. The odour tests of cherriesin this research include from this research were carried out by twelve trained panel members in three replicates using a sensory scale with increasing points from 1 to 10 points regarding the intensity of the sensory characteristics. The aroma compounds occurring in apples with the highest content of volatiles were separated and identified using combined GC-MS. Four members of the sensory panel described on shift the odour from the GC column answering the question yes or no regarding mouldy, mutzu earthy odour. The statistical methods included one-way and multiple analysis of variance using a Statgraphic Statistical Package (Statistical Graphics, Version 4, Rockville, USA). Averages were separated using letters. Averages were separated by letters.

		Brine							
Drained weight, g		Soluble solids, g 100 g ⁻¹		Acidity g 100 g ⁻¹		Soluble solids, g 100 g ⁻¹		Acidity g 100 g ⁻¹	
prick-	non-	prick-	non-	prick-	non-	prick-	non-	prick-	non-
led	prickled	led	prickled	led	prickled	led	prickled	led	prickled
289a	215d	20.0a	19.7a	1.13a	1.16a	19.3b	19.8b	1.09a	1.11a
299b	225c	20.8a	20.2a	1.16a	1.15a	20.4a	20.5a	1.12a	1.10a
297b	256b	19.5a	20.3a	1.09a	1.07a	19.7a	19.9a	1.06a	1.11a
308a	256b	20.9a	20.2a	1.15a	1.14a	20.6a	20.3a	1.11a	1.10a
307a	268a	19.9a	19.8a	1.12a	1.10a	20.3a	20.3a	1.08a	1.06a
300a	244b	20.2a	20.0a	1.10a	1.10a	20.1a	20.2a	1.09a	1.10a

Table 1. Effects of prickling on drained weight, and brine compounds by blanching for 6 min at 85 °C.

3. RESULTS

Drained weight of pickled cucumber increased significantly with storage time up to 114 days increased up to storage for 187 days (Table 1). Drained weight were significantly highest for pickled cucumber and increased significantly with storage time up to 114 days whereas drained weight of non-pickled cucumber were significantly lowest for non-pickled cucumber and increased up to storage for 187 days. The contents of soluble solids and acidity were non-significantly different for pickled and non-pickled cucumbers (Table 1). Drained weight increased significantly with days of storage time and up to blanching for 10 min by blanching at 72 °C and to 12 min by blanching at 85 °C (Table 2). Drained weight was significantly higher by blanching at 85 °C in comparison to blanching at 72 °C. The reason for this was a higher increase in drained weight at 85°C that was 30 g in comparison to 12 g by blanching at 72°C.A significantly low firmness was obtained by blanching time 0 and 3 min and a maximum occurred after storage for 176 days, and thereafter firmness decreased significantly in comparison to firmness after 176 days (Table 3). After blanching for 6 min were the firmness significantly lowest after 91 days on storage and significantly increases was found for the remaining part of the storage period. The pattern of data from blanching at 85°C for 0 min was almost equal to the data obtained by blanching for 0 min at 72° C. However, the firmness at all storage days was significantly highest by further blanching of the cucumber at 85°C. This pattern included a significant higher firmness after two-step blanching that is due to pectinolytic and cellulytic enzyme activities. The explanation is that the increases in firmness are related to hydrolysis of pectins and cellulose into sugar acids with a large number of free carboxylic acid groups. These carboxylic acids are recombined by divalent mineral ions such as Ca⁺⁺and Mg⁺⁺[1,2, 3,as described previously for firming of potatoes and green beans [5, 7]. Further increases in firmness occurred by increase of the blanching temperature from 72 to 85°C (Table 3). The increases in firmness during blanching at 85°C resulted in maximum firmness 397, 387, 380, 367 and 361 kg for 0, 3, 6, 9 and 12 min that may be compared with 376, 383, 377, 375, and 384 kg found for blanching at 72°C. On the basis of these data it was concluded that further blanching at 85°C resulted in further increases in firmness and the decreases occurred by blanching for 9 and 12 min. The temperature in the middle of two cucumbers with diameter 2.0 and 2.9 cm reached 72 °C after 13.5 and 16.5 min, respectively, which may be considered as sufficient blanching time at these temperatures.Data from evaluation of the effect of blanching of cucumber slices at 80 °C previously has shown that browning may be a serious problem using short blanching times and by storage of cucumber slices and whole pickled cucumber at temperatures above 5 °C [7]. An experiment with storage of pickled cucumber showed that drained weight increased significantly with storage time, while firmness decreased significantly with storage time. Storage of pickled cucumber at 12 and 20 °C did not result in significance. An experiment with blanching of pickled cucumber for 0, 3, 6, 10 and 15 min and storage at 12 and 20 °C showed that drained weight increased significantly with blanching time and that firmness decreased with increases in the storage temperature (Table 4).

Table 2. Drained weight of cucumber blanched at 72 and 85 °C for 0 to 12 min and cool storage at 12 °C (n =20).

Time	Blanching at 72 °C, min					Blanching at 85 °C, min				
Days	0	3	6	9	12	0	3	6	9	12
91	294d	302c	308b	311a	310a	284e	302d	306c	313b	319a
176	306b	305b	309b	314a	315a	292e	306d	312c	317b	321a
272	307a	310a	316a	316a	314a	293d	311c	316b	320a	322a
327	301c	310a	314a	315a	314a	294d	309c	314b	316b	320a
Average	302c	307b	312a	314a	313a	291e	307d	312c	317b	321a

Table 3. Firmness of sour-sweet cucumber blanched at 72 and 85 °C for 0 to 12 min ($n = 20$).										
Days	Blanching at 72 °C, min						Blanch	ing at 85 °	C, min	
	0	3	6	10	0	3	6	9		

Days		Blanch	ing at 72 °	C, min		Blanching at 85 °C, min				
	0	3	6	10	12	0	3	6	9	12
91	362c	374b	367b	371b	365c	389b	377c	366c	356c	350c
176	376a	383a	375a	375a	370b	397a	385a	377a	363b	357b
272	369b	369c	376a	371b	384a	385c	381b	380a	367a	356b
327	363c	369c	377a	371b	381a	382c	387a	374b	356c	361a

Table 4. January. Sensory properties after storage for 7 months.

Blanching at 80°C, min	0	3	6	9
Browning, row no (1 to 5)	3.3a	2.5b	2.5c	1.8d
Firmness, row no (1 to 10)	7.1a	7.0a	5.6b	5.6b
Pasteurization at 75 °C, min	15	30	45	60
Browning, point (1 to 5)	2.5a	2.5a	2.6a	2.4b
Firmness, row no (1 to10)	6.6b	7.7a	7.4a	6.3b
Storage temperature, °C	5	10	15	20
Browning after 7 months (1 to 5)	1.0a	2.0b	3.0c	4.0d
Firmness after 7 months (1 to10)	8.4a	7.6a	6.0b	5.8b
Ascorbic acid, mg 1.2 kg ⁻¹	0	83	166	333
Browning (1 to 5)	1.5d	2.5c	3.0b	3.8a
Firmness, point	6.3a	5.9b	5.4c	5.1d
Sensory evaluation, point $1 =$ lowest, whereas 5 and 1				

4. DISCUSSION

According to table 1 resulted prickling in a considerable and significantly increase in drained weight that in comparison with non-pickled cucumber may cause a more delicious texture and flavour [5]. The data from determination of soluble solids and acetic acid showed that an almost complete equilibrium was obtained for soluble solids an acetic acid in pickled and non-pickled cucumber, whereas the

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drained weight of pickled cucumber were significantly higher if the cucumber are pickled in comparison to non-pickled cucumber (Table 1). Drained weight increased significantly with treatment time at both blanching temperatures and the data shows that a higher blanching temperature result in a higher drained weight (Table 2). Equilibration between whole cucumbers and the brine was found to be consistent with a controlled first-order rate process for attainment of equilibrium for sugar and malic acid from the cucumbers to the brine and by NaCl and acetic acid [6]. The rates of equilibrium were much higher for peeled cucumbers in comparison to the equilibration rate for non-peeled fruits and they increased by raising temperatures for sugar, malic acid and NaCl [6]. Attainment of equilibrium for solutes was linearly related to time. Significantly differences changes was obtained during blanching for 10 and 15 minutes at the temperature 72 °C, while equality by blanching at 85 °C was obtained after the longest treatment time 12 min. The change in firmness by blanching at 85 °C decreases on average with blanching time (Table 3), which is guite clear regarding the averages. Based on these data it was concluded that firmness decreased significantly with blanching time. Another conclusion may be that there is a significant difference regarding blanching at the two temperatures because of the changes in firmness that may be considered as very satisfying to use 72 °C because a lower blanching temperature may result in lower decreases in dry matter that may be considered as very valuable and reduce the cost by processing. According to the data presented above increases the firmness very little by blanching at 72 °C with firmness f = 370 + 0.3kg and using higher temperatures resulted in less firm cucumber with f = 390-2.88min.

	Drained	weight, g	Firmness, kg				
Min	72 °C	85 °C	Min	72 °C	85 °C		
0	0	291	0	368	388		
3	307	307	3	374	377		
6	312	312	6	374	374		
10	314	317	10	371	356		
15	313	321	15	375	361		

Table 5. Drained weight and firmness at 72 °C and 85 °C.

Cucumbers contain pectin esterase that may de-methylate pectins in the cell walls and cause serious softening of the cucumber tissue during storage if they not are blanched sufficiently [2]. In some cases must softening be avoided or require use of calcium chloride that are connecting the pectin molecules together and make the cucumber products qualified because of a properly tissue strength [8]. An example could be blanching of cucumber with a low content of calcium that may need extra heating in order to inactivate pectinesterase and decrease the content of glycosides that are present in cucumber [9].Precondition conditioning of cucumber tissues by pre-blanching may result in improvement of cucumber firmness and retention of a white appearance of cucumber tissues if they are processed within one hour or within 24-28 hours after pre-condition[10]. Firmness decreased by blanching from 16.5 N to 7.0 N for 12 hours using pectinesterase and glucosidase [11]. During storage increased firmness losses with increasing temperature and pH and firmness decreased with increasing content of salt due to interaction between temperature and pH [12]. Blanchingmay result in improvement of cucumber firmness and improve the white appearance of cucumber [10]. Improvement of firmness may be attributed to formation of calcium pectate that improves the middle lamella cell wall rigidity [8]. Carbohydrates, water soluble pectins, and sodium hexametaphosphate-soluble pectin were significantly correlated with shear press and sensory firmness. Cucumber tissues may be cured using knowledge about rates of cucumber tissue including the effects of pH, calcium, and temperature [8, 9]. Softening of cucumber tissue may occur because of a low content of calcium and may be reduced using calcium because demethylation of fresh Ca-pectates that improves cell wall rigidity and improved resistances against polygalacturonase [13]. The relation between polygalacturonase was hyperbolic and that means that log10 versus time is a straight line [14]. A low content of Calcium may result in a lower firmness. Pectic substances of fresh cucumber tissue are from 35-50% deesterefied and the free acid groups may react with minerals such as calcium Ca++ and Mg++ and that result in strong connections between various pectin molecules that increase the number of bindings between various pectins in cucumber. A low content of calcium may result in low firmness [11]. Both calcium and NaCl prevented both enzymatic [14] and non-enzymatic softening of cucumber [8]. Polyphenol oxidase (Tyrosinase) catalyses hydroxylation of monophenols and promote oxidation of the O-diphenoles to O-quinones that may cause severe browning of cucumber tissue [15]. Enzymatic browning of fruits and vegetables is therefore mostly related to oxidation of endogenous phenolic compounds that ultimately leads to the formation of dark brown polymers of a quinoidal nature. That situation requires blanching of cucumbers in order to avoid a serious loss in consumer qualities. Peroxidase constitutes a group of enzymes (glycoproteins) whose main function is oxidation of different substrates at the expense of $H_2O_2[2]$. These enzymes are also involved in lignification, hormone metabolism, and response to stress in plant tissues. It is also necessary to inactivate the thermo-stable peroxidase because this enzyme may cause adverse flavour and colour in short time after processing. A test for presence or absence of peroxidase is necessary in order to avoid severe development of various aroma compounds in processed fruits and vegetables [12]. Ascorbic acid induced browning of (+)-catechin in a model wine system has been studied. A significant increase in absorbance at 440 nm was observed over 14 days when ascorbic acid was incubated at 45 °C with (+)-catechin in a model wine base. The onset of browning was delayed for about 2 days, although the length of the lag period was dependent on the amount of molecular oxygen in the headspace of the reaction system. The lag period was not observed when a pre-oxidized solution of ascorbic acid was used, suggesting that a product of ascorbic acid oxidation is responsible for the onset of browning [13]. Ascorbic acid was a very efficient antioxidant against formation of hexanal, (E)-2-penenal, (E)-2-hexenal, (E)-2-heptenal, and (E)-2octenal when slurries of fermented cucumber tissue were exposed to oxygen and most of these oxidative was efficiently prevented at 1755 ppm [14].Non-chemical browning of cucumber includes interactions of amino acids and sugars at various levels of pH that could affect the colour of cucumber by non-enzymatic browning through several kinetic reactions that would reduce the contents of amino acids and sugars [11, 20]. Model experiments showed that the level of 5-hydroxymethylfurfural (5hmf) at pH 3.5 from fructose was 31.2 times faster than glucose, whereas sucrose was 18.5 times faster than glucose [21]. The contents of citric acid, calcium, magnesium, potassium, and the amino acids alanine, aspartic acid, and y-aminobutyric acid could accelerate formation of 5hydroxylfumerate.Model systems were employed to determine the relative reactivity of sucrose, glucose, and fructose in the formation of 5-hydroxymethylfurfural (5-hmp) at pH 3.5. Fructose was the major reactant for formation of 5-HMF because this carbohydrate was 31.2 times faster than glucose, while sucrose was faster than glucose in the rate of 5-HMF formation when averaged over three different sugar-catalyst systems[22]. Accelerating effects of citric acid, minerals (calcium, magnesium, and potassium), and amino acids (alanine, aspartic acid, and γ -aminobutyric acid) in the formation of 5-HMF from sugars were evaluated. With fructose as the substrate was 5-HMF produced 5 times faster by presence of citric acid and minerals than in the presence of HCl. Varying catalytic effects were noted with the three amino acids. Formation of 5-HMF formation from glucose and sucrose was enhanced by presence of the amino acids, whereas virtually no enhancement occurred when fructose was the substrate [23].Large whole cucumbers (size no. 4) were fermented and held in solutions containing low NaCl with and without CaCl₂, at 24°C for one and four 4 months. Firmness of pericarp tissues (centre) of pickles exposed to CaCl₂, was much greater than of pickles not exposed to CaCl₂, after 1 and 4 months in storage. Locular tissue of pickles not exposed to CaCl₂, softened dramatically between 1 and 4 months in storage and after 4 months were the centres of cucumber extremely soft. In contrast, centre of pickles exposed to CaCl₂, were firm and intact after 1 and 4 months. Addition of CaCl₂, to fermentation and storage brines appears to provide a simple method for pickle improvement. Pasteurization of pickle products for 15 min until 73 °C in the jar centre was adequate to avoid spoiling [24] and pasteurizing for 25 min at 82 °C was enough time to reduce the level of microorganism and peroxidase to an acceptable level [25], whereas pasteurizing of pickled cucumber at 75, 80, 85 and 90 °C for 22.0, 19.5, 14.5 and 4 min, indicated that an residual activity of polygalacturonase could be expected in the pasteurized pickle products [26, 27].

Pasteurization of cucumber at internal product temperature 71-76 °C followed by prompt cooling may be considered as efficient to avoid spoilage and result in a high quality of pickled cucumber [28]. Optimum degree of esterification in order to obtain an optimum firmness was between 10 and 12 newton[29].Firmness of cucumber pasteurized at 74 °C was spoiled after 120, 30 and 30 min using water, acetic acid and lactic acid in the brine at 74 °C, respectively [30]. Acetic acid caused the least loss in firmness, followed by lactic, citric, malic and oxalic acids [30] and peroxidase activity in cucumber prickles decreased significantly during the first four days in cucumber without and with CaCl₂, while peroxidase in the brine increased during the first week and decreased during the second week, respectively. Firmness decreased with increased by rising temperature and pH, whereas firmness increased with increases in salt and the interaction between temperature and pH [31]. The damage on texture may be severe 20 °C and especially by higher temperature.

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